# (12) UK Patent Application (19) GB (11)

## 2 020 424 A

- (21) Application No 7914580
- (22) Date of filing 26 Apr 1979
- (23) Claims filed 26 Apr 1979
- (30) Priority data
- (31) 903109
- (32)5 May 1978
- (33)**United States of America** (US)
- Application published (43)14 Nov 1979
- (51) INT CL2
- G01N 33/16
- Domestic classification G1B BA BR
- (56) Documents cited None
- (58) Field of search
- G1B
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#### (54) Determination of glucose

(57) An improved hexokinase coupled enzyme assay procedure, for measuring the glucose content of body fluids, comprises addition of D-Mannose to the reaction system.

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#### **SPECIFICATION**

#### Kinetic measurement of glucose concentrations in body fluids and formulations for use therein

5 This invention relates to a novel method and formulation for the kinetic determination of glucose concentration in body fluids using the hexokinase coupled enzyme assay method.

The know method of coupling the enzymes hexokinase and glucose-6-phosphate (G6P) dehydrogenase offers a highly specific system for the determination of glucose in body fluids, such as serum, by reacting the sample containing glucose to completion according to the following enzymatic steps (Adenosine diphosphate is ADP; adenosine triphosphate is ATP; Nicotinamide-adenine-dinucleotide is NAD; reduced

NAD is NADH):

(1)

15

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(2)

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G6P dehydrogenase

G6P + NAD 6-phosphogluconate + NADH

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The formation of NADH, measured spectrophoto-metrically is 340 nm, is directly proportional to the glucose concentration in the fluid sample once all the glucose has been reacted.

It has been recognised that a need exists for an assay procedure that eliminates the necessity for completion of the enzyme coupling reaction in order to give greater sample through-put/unit time. It has also been recognised as desirable to eliminate the need for serum blank absorbance corrections which are costly and time-consuming. Moreover, it would be desirable to have a kinetic reaction rate assay procedure which was generally more specific and more easily automated.

It has heretofore been proposed to employ a kinetic fixed time assay for glucose using the hexokinase/glucose-6-phosphate dehydrogenase reagent system employing N-acetyl-glucosamine as a 30 selected reaction kinetic inhibitor.

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It has now been found that a suitable coupled enzymatic assay procedure using the hexokinase/glucose-6-phosphate dehydrogenase reagent system can be employed for the kinetic measuring of a body fluid for glucose concentration where the rate of NADH formation is proportional to the rate of glucose reacting, which in turn is proportional to the glucose concentration in the body fluid sample.

Such an improved coupled enzyme assay procedure has been made possible by the finding that the addition of D-mannose alters the kinetic properties of the hexokinase enzyme reaction step. The D-mannose is converted to mannose-6-phosphate *in situ* which acts as a product inhibitor of hexokinase thereby causing the rate of the NADH formation to be proportional to the glucose concentration in the sample.

The resulting system exhibits linear kinetics at about 30°C. for at least one to more than five minutes, 40 depending upon glucose concentration.

It has generally been found that the inclusion of about 4 grams of D-mannose per liter of reagent accomplishes the desired result of providing a system whereby the rate of NADH formation is proportional to the rate of glucose concentration in the fluid sample. Such a kinetic system of glucose determination compares very favorably in accuracy and precision with the heretofore used glucose end-point method.

45 Thus, this kinetic system is adaptable for use on any centrifugal or fast kinetic analyzer, such as, for example, the Gilford 2000 or 3500 models, the Perkin-Elmer KA-150 and other analyzers.

The kinetic reaction system of this invention provides a linear calibration curve over the glucose concentration range of from about 50 to at least about 750 mg/dl and is linear for a period of up to at least 5 minutes over a temperature range of about 23 to 37° C.

50 The rate at which glucose is reacting is determined by measuring the absorption of NADH spectrophotometrically at about 334 to 366 nm, preferably at 340 nm.

Reagent formulations for the carrying out of the kinetic reaction measurement of glucose concentration according to this invention and depending upon the nature of the clinical instrumentation on which the reagent is meant to be used can be formulated by the inclusion of D-mannose in either a single-vial

55 formulation wherein all the necessary constituents for the test reaction are included in one sample container or in a dual-vial formulation wherein one vial contains a "substrate" containing all reactants except one and a second vial containing the "trigger solution".

Typical examples of such formulations and their preparation are set forth in the following examples:

Example 1				
Single Vial Formulation				

Sin	gle Vial Formulation				
5	Ingredient	Molecular Weight .	Specific Amount	Range of Amounts	5
1.	Add:				•
	Dextran	75,000	30 g/l	15-60 g/l	10
10	Bovine Serum Albumin	••	4 g/l	2-8 g/l	10
15	PIPES Buffer Pipera- zine - N, N'-bis (2- ethanesulfonic acid), monosodium, monohydrate	324	100 mmoles	80-120 mmoles	15
20	Magnesium Chloride, Hexahydrate	203	32 mmoles	3034 mmoles	20
20	Adenosine-5'-triphos- phate, disodium tri- hydrate (ATP)	605	25.2 mmoles	24.226.2 mmoles	
25	Nicotinamide-adenine- dinucleotide, tri- hydrate (NAD)	717	3.8 mmoles	2.5–5.0 mmoles	25
30	D-Mannose, anhydrous	180	24.0 mmoles	23.0-25.0 mmoles	30
35 2.	To: 800 ml Deionized Water. Stir until com- pletely dissolved.  Adjust pH to 7.0 with 50% Sodium Hydroxide Solution at about 30° C.				35
40 3.	Adjust volume to 1 liter with deionized water. Cool mixture to about 1 to 8° C.				40
45 4.	Add:				45
<b>50</b> .	Glucose-6-phosphate de- hydrogenase, ex. L. mesenteroides Hexokinase, ex. yeast		6,000 IU (@ 37°C.) 12,000 IU (@ 37° C.)	5,000 7,000 iU's 10,00014,000 IU's	50
5. 55	Dispense 5.0 ml of so- lution into 30 cc serum bottles.				55
6.	Freeze to about -30° C.				٠
60 7.	Lyophilize about 36 to 48 hours with +25° C. shelf heat applied as soon as 200 micron vacuum is reached.				60
65	Final product temperature not to exceed +30° C.				65

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	State				
8.	Stopper and seal under vacuum. Store at 2 to 8°C.				
9.	Each vial so prepared is to				
5	be reconstituted with 20.0 ml distilled or delonized				5
	mi distilled or delonized water for use in reaction				
	test system.				
Exa	ample II				
10	•				10
Du	al Vial Formulation				
		Molecular	Specific	Range	
	Ingredient	Weight	Amount	of Amounts	
15 1.	For SUBSTRATE, add:				15
١.	rui SUBSTRATE, aug.				
	Dextran	75,000	30 g/l	15-60 g/l	
20	Bovine Serum Albumen		3.3 g/l	2-8 g/l	20
	PIPES Buffer Pipera-				
	zine - N, N'-bis (2-				
05	ethanesulfonic acid),				25
25	monosodium, mono- hydrate	324	82.5 mmoles	66-99 mmoles	20
	.,,,,,,,,				
	Magnesium Chloride,				
••	Hexahydrate	203	26 mmoles	24–28 mmoles	30
30	Adenosine-5'-triphos-				
	phate, disodium tri-				
	hydrate (ATP)	605	2.2 mmoles	2.1-2.3 mmoles	
35	Nicotinamide-adenine-				35
00	dinucleotide, tri-				
	hydrate (NAD)	717	3.14 mmoles	2.5-5 mmoles	
	D-Mannose, anhydrous	180	22.0 mmoles	21.4-22.6 mmoles	
40	D-Maimoso, amyurous	100	LL.O HITTOTOS	LIIT LEW HILLOGS	40
	To: 800 ml Deionized				
	Water. Stir until				
	completely dissolved.				
45 2.	Adjust pH to 7.0 with				45
	50% Sodium Hydroxide So-				
	lution at about 30° C.				
3.	Adjust volume to 1 liter				
50	with deionized water.				50
	Cool mixture to about				
	1 to 8° C.				
4.	Add:				
55	Glugges & sheeshets				55
	Glucose-6-phosphate dehydrogenase, ex. L.		6,000 IU	5,000- 7,000	
	mesenteroides		(@ 37°C.)	IU's	
60	Havakinasa ay yanat		12,000 IU	10,000-14,000	60
60	Hexokinase, ex. yeast	-	12,00010 (@ 37°C.)	10,000 14,000 IU's	-
	<b>.</b>		,		
5.	Dispense 6.0 ml of so- lution into 30 cc				
65	serum bottles.				65

serum bottles.

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6.	Freeze to about30°C.				
7.	Lyophilize about 36 to				
	48 hours with +25°C.				_
5	shelf heat applied as				5
	soon as 200 micron				
	vacuum is reached.				
	Final product tempera-				
	ture not to exceed +30°C.				10
10	Ctanana and analysis day				10
8.	Stopper and seal under vacuum. Store at 2 to	•			
	8°C.				
	<b>6 C.</b>				
15 9.	Each vial so prepared is				15
	to be reconstituted with				
	20.0 ml distilled or de-				
	ionized water for use in				
	test.				
20					20
10.	For TRIGGER, add:				
	D. 1	75.000	20 ~//	10 25 -8	
	Dextran	75,000	20 g/l	10-25 g/i	
<b>2</b> 5	PIPES Buffer	324	16.2 mmoles	14–18 mmoles	25
	ATP	605	172 mmoles	168-176 mmoles	
11.	Adjust pH to 7.0 (30°C)				
30	with 50% Sodium Hy-				30
	droxide Solution at				
	about 30°C.				
				•	
12.	Adjust volume to 1				35
35	liter with deionized				33
	water.				
40	Diameter 4.0 mil of an				
13.	Dispense 4.0 ml of so-				
40	lution into 15 cc				40
40	serum bottles.	•			
14.	Freeze to about -30°C.				
14.	Treeze to about 50 c.				
15.	Lyophilize about 36 to				
45	48 hours with +25°C.				45
	shelf heat applied as				
	soon as 200 micron				
	vacuum is reached.				
	Final product temper-			•	
50	ature not to exceed				50
	+30°C.				
16.	Stopper and seal under				
101	vacuum. Store at 2 to				
55	8°C.				55
JJ	<del>.</del> .				
17.	Each vial so prepared				
	is to be reconstituted				
	with 11 0 ml distilled				

with 11.0 ml distilled

use in test.

or deionized water for

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60

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used instead other binders such as gum arabic, mannitol, sorbitol or the like. As other suitable protein sources in place of bovine serum albumin there may be employed, globulins, gelatins or the like. Other suitable buffers include, for example, tris(hydroxymethyl)amino methane, triethanol amine, N-2-hydroxyethylpiperazine-N'-2-ethane sulfonic acid and the like. Other suitable magnesium salts would include magnesium acetate, nitrate, sulfate and the like.

The usefulness of the formulations of this invention and the linearity of the kinetic reaction rate of the change in absorption of NADH is shown by the following Examples wherein the single vial formulation of Example I was used to measure the change in absorption at 340 nm at 30°C. for standard samples of known glucose concentration. The change in absorption was determined by utilizing a Gilford model 2000 spectrophotometer.

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#### Example III

#### Assay volume, 2.0 ml reagent + 10 µL sample

	Standard Glucose Concentration mg/dl	Observed rate Δ A 340 nm/min	Duration of Linear Rate, minutes	15
	100	.007	5	
20	200	.014	3	20
	300	.022	3	
25	500	.034	2-1/2	25
	750	.047	2-1/2	
	1000	.060	2-1/2	00
30				30

#### Example IV

## Assay volume, 2.0 ml reagent + 20 µL sample

35	Standard Glucose Concentration mg/dl	Observed rate $\triangle$ A 340 nm/min.	Duration of Linear Rate, minutes	35
	100	.014	4	4.5
40	200	.028	3	40
	300	.040	3	
45	500	.065	2-1/2	45
	750	.081	2	
	1000	.095	2	

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### Example V

Assay volume, 2.0 ml + 25 µL sample

5	Standard Glucose Concentration, mg/dl	Observed rate Δ A 340 nm/min.	Duration of Linear Rate, minutes	5
	100	.019	4	
10	200	.035	2-1/2	10
	300	.051	2-1/2	
	500	.066	2-1/2	15
15	750	.086	2-1/2	,,
	1000	.102	2	
20	CLAIMS			20
	<ol> <li>A method for the determination of glucose concentration in a fluid sample by the hexokinase coupled enzyme assay procedure, in which D-mannose is added to the enzyme reaction reagent system, so that the 25 overall reaction rate is substantially proportional to the glucose content of the fluid sample.         <ol> <li>A method according to claim 1 in which the glucose concentration is determined by measuring the absorption of nicotinamide-adenine-dinucleotide reduced at 340 nm.</li> <li>A method according to claim 1 or claim 2 in which the amount of D-mannose in the reagent system is about 6 millimoles/litre of reagent.</li> </ol> </li> <li>A method according to claim 1 substantially as described in Example III.</li> <li>A reagent formulation for the determination of glucose concentration in a fluid sample by the</li> </ol>			
	hexokinase coupled enzyme assay 6. A formulation according to cl reagent.	procedure, the formulation comprisir aim 5 in which the amount of D-manr	ng D-mannose. nose is about 6 millimoles/litre of	or
35	7. A formulation according to cl	aim 5 substantially as described in Ex	rample I or Example II.	35

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